

### MATERIAL SAFETY DATA SHEET

| Section 1: Chemical Product and Company Identification   |
|--|
| Product Name: Sodium bicarbonate   |
| Catalog Codes: SLS3241, SLS2446, SLS3868   |
| CAS#: 144-55-8   |
| RTECS: VZ0950000   |
| TSCA: TSCA 8(b) inventory: Sodium bicarbonate  |
| Cl#: Not available.  |
| Synonym: Baking Soda; Bicarbonate of soda; Sodium acid   |
| carbonate; Monosodium carbonate; Sodium hydrogen   |
| carbonate; Carbonic acid monosodium salt   |
| Chemical Name: Sodium Bicarbonate  |
| Chemical Formula: NaHCO3   |
| Contact Information:   |
| Company: Aurora Industry Co.,Ltd.  |
| Address:Room 7033, No.9-1, Haifu Road, Dalian Free Trade Zone, China                           |
| Tel:0411-82288674  |
| Section 2: Composition and Information on Ingredients  |
| Composition:   |
| Name CAS # % by Weight   |
| Sodium bicarbonate 144-55-8 100  |
| Toxicological Data on Ingredients: Not applicable.   |
| Section 3: Hazards Identification  |
| Potential Acute Health Effects: Slightly hazardous in case of skin contact (irritant), of eye  |
| contact (irritant), of ingestion, of   |
| inhalation.  |
| Potential Chronic Health Effects:  |
| CARCINOGENIC EFFECTS: Not available.   |
| MUTAGENIC EFFECTS: Not available.  |
| TERATOGENIC EFFECTS: Not available.  |
| DEVELOPMENTAL TOXICITY: Not available.   |
| Repeated or prolonged exposure is not known to aggravate medical condition.                    |
| Section 4: First Aid Measures  |
| Eye Contact:   |
| Check for and remove any contact lenses. In case of contact, immediately flush eyes with       |
| plenty of water for at   |
| least 15 minutes. Cold water may be used. Get medical attention if irritation occurs.          |
| Skin Contact:  |
| Wash with soap and water. Cover the irritated skin with an emollient. Get medical attention if |
| irritation develops.   |



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Cold water may be used.

Serious Skin Contact: Not available.

#### Inhalation:

If inhaled, remove to fresh air. If not breathing, give artificial respiration. If breathing is difficult,

give oxygen. Get

medical attention.

Serious Inhalation: Not available.

#### Ingestion:

Do NOT induce vomiting unless directed to do so by medical personnel. Never give anything by mouth to an

unconscious person. Loosen tight clothing such as a collar, tie, belt or waistband. Get medical attention if

symptoms appear.

Serious Ingestion: Not available.

#### **Section 5: Fire and Explosion Data**

Flammability of the Product: Non-flammable.

Auto-Ignition Temperature: Not applicable.

Flash Points: Not applicable.

Flammable Limits: Not applicable.

Products of Combustion: Not available.

Fire Hazards in Presence of Various Substances: Not applicable.

Explosion Hazards in Presence of Various Substances:

Risks of explosion of the product in presence of mechanical impact: Not available.

Risks of explosion of the product in presence of static discharge: Not available.

Fire Fighting Media and Instructions: Not applicable.

**Special Remarks on Fire Hazards:** When heated to decomposition it emits acrid smoke and irritating fumes.

Special Remarks on Explosion Hazards: Not available.

#### **Section 6: Accidental Release Measures**

Small Spill:

Use appropriate tools to put the spilled solid in a convenient waste disposal container. Finish cleaning by

spreading water on the contaminated surface and dispose of according to local and regional authority

requirements.

#### Large Spill:

Use a shovel to put the material into a convenient waste disposal container. Finish cleaning by spreading water

on the contaminated surface and allow to evacuate through the sanitary system.

#### Section 7: Handling and Storage

#### **Precautions:**

Do not ingest. Do not breathe dust. If ingested, seek medical advice immediately and show the



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container or the

label. Keep away from incompatibles such as acids.

**Storage:** Keep container tightly closed. Keep container in a cool, well-ventilated area.

#### **Section 8: Exposure Controls/Personal Protection**

#### **Engineering Controls:**

Use process enclosures, local exhaust ventilation, or other engineering controls to keep airborne levels below

recommended exposure limits. If user operations generate dust, fume or mist, use ventilation to keep exposure to

airborne contaminants below the exposure limit.

**Personal Protection:** Safety glasses. Lab coat. Dust respirator. Be sure to use an approved/certified respirator or equivalent.

Gloves.

#### Personal Protection in Case of a Large Spill:

Splash goggles. Full suit. Dust respirator. Boots. Gloves. A self contained breathing apparatus should be used

to avoid inhalation of the product. Suggested protective clothing might not be sufficient; consult a specialist

BEFORE handling this product.

Exposure Limits: Not available.

#### **Section 9: Physical and Chemical Properties**

Physical state and appearance: Solid.

Odor: Odorless.

Taste: Saline. Alkaline.

Molecular Weight: 84.01g/mole

Color: White.

pH (1% soln/water): Not available.

Boiling Point: Not available.

Melting Point: Not available.

Critical Temperature: Not available.

Specific Gravity: Density: 2.159 (Water = 1)

Vapor Pressure: Not applicable.

Vapor Density: Not available.

Volatility: Not available.

Odor Threshold: Not available.

Water/Oil Dist. Coeff.: Not available.

lonicity (in Water): Not available.

Dispersion Properties: See solubility in water.

#### Solubility:

Soluble in cold water.

Slightly soluble in alcohol.

Solubility in Water: 6.4, 7.6, 8.7, 10.0, 11.3, 12.7, 14.2, 16.5, 19.1 g/100 solution at 0, 10, 20,

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30, 40, 50, 60, 80, adn 100 deg. C, respectively. Solubility in Water: 6.9, 8,2, 9.6, 11.1, 12.7, 14.5, 16.5, 19.7, and 23.6 g/100g water at 0, 10, 20, 30, 40, 50, 60, 80, 100 deg. C, respectively. Section 10: Stability and Reactivity Data Stability: The product is stable. Instability Temperature: Not available. Conditions of Instability: Incompatible materials, Moisture. Stable in dry air, but slowly decomposes in moist air. Incompatibility with various substances: Reactive with acids. Corrosivity: Non-corrosive in presence of glass. **Special Remarks on Reactivity:** Reacts with acids to form carbon dioxide. Dangerous reaction with monoammonium phosphate or a sodium-potassium alloy. Special Remarks on Corrosivity: Not available. Polymerization: Will not occur. Section 11: Toxicological Information Routes of Entry: Inhalation. Ingestion. Toxicity to Animals: Acute oral toxicity (LD50): 3360 mg/kg [Mouse]. Chronic Effects on Humans: Not available. Other Toxic Effects on Humans: Slightly hazardous in case of skin contact (irritant), of ingestion, of inhalation. Special Remarks on Toxicity to Animals: Not available. **Special Remarks on Chronic Effects on Humans:** Sodium Bicarbonate as produced genetic effects in rats (unscheduled DNA synthesis). However, no affects have been found in humans. Special Remarks on other Toxic Effects on Humans: Acute Potential Health Effects: Skin: May cause mild skin irritation. Eyes: May cause mild eye irritation. Inhalation: May cause respiratory tract irritation. Symptoms may include coughing and Ingestion: Symptoms of overexposure to Sodium Bicarbonate include thirst, abdominal pain, gastroenteritis, and inflammation of the digestive tract. **Chronic Potential Health Effects:** Skin: Repeated or prolonged skin contact may cause irritation, drying or cracking of the skin. Ingestion and Inhalation: Chronic toxicity usually occurs within 4 to 10 days following ingestion of very large amounts. Repeated or prolonged ingestion or inhalation of large amounts may cause

sneezing.



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metabolic abnormalities, and

sodium retention. Metabolic abnormalities such as acidosis, hypernatremia, hypochloremia, alkalosis,

hypocalcemia, or sodium retention may affect the blood, kidneys, respiration (cyanosis, apnea secondary to

metabolic acidosis or pulmonary edema), and cardiovascular system (tachycardia,

hypotension).

Severe toxicity may also affect behavior/central nervous system/nervous system. Neurological changes may

result from metabolic abnormalities. These may include fatigue, irritability, dizziness, mental confusion,

paresthesia, seizures, tetany, cerebral edema

Medical Conditions Aggravated by Exposure: Persons with pre-existing skin conditions might have increased

sensitivity. Predisposing conditions that contribute to a mild alkali syndrome include, renal disease, dehydration,

adn electrolyte imbalance, hypertension, sarcoidosis, congestive heart failure, edema, or other sodium retaining

conditions.

#### Section 12: Ecological Information

Ecotoxicity: Not available.

BOD5 and COD: Not available.

Products of Biodegradation:

Possibly hazardous short term degradation products are not likely. However, long term degradation products may

arise.

**Toxicity of the Products of Biodegradation:** The product itself and its products of degradation are not toxic.

Special Remarks on the Products of Biodegradation: Not available.

#### Section 13: Disposal Considerations

Waste Disposal:

Waste must be disposed of in accordance with federal, state and local environmental control regulations.

#### **Section 14: Transport Information**

**DOT Classification:** Not a DOT controlled material (United States).

Identification: Not applicable.

Special Provisions for Transport: Not applicable.

#### Section 15: Other Regulatory Information

Federal and State Regulations: TSCA 8(b) inventory: Sodium bicarbonate

Other Regulations: Not available.

Other Classifications:

WHMIS (Canada): Not controlled under WHMIS (Canada).



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### DSCL (EEC): This product is not classified according to the EU regulations. Not applicable. HMIS (U.S.A.): Health Hazard: 1 Fire Hazard: 0 Reactivity: 0 **Personal Protection:** E National Fire Protection Association (U.S.A.): Health: 1 Flammability: 0 Reactivity: 0 Specific hazard: **Protective Equipment:** Gloves. Lab coat. Dust respirator. Be sure to use an approved/certified respirator or equivalent. Safety glasses. Section 16: Other Information References: Not available. Other Special Considerations: Not available. Created: 24/05/2020 10:02 PM The information above is believed to be accurate and represents the best information currently available to us. However, we make no warranty of merchantability or any other warranty, express or implied, with respect to such information, and we assume no liability resulting from its use. Users should make their own investigations to determine the suitability of the information for their particular purposes. In no event shall ScienceLab.com be liable for any claims, losses, or damages of any third party or for lost profits or any special, indirect, incidental, consequential or exemplary

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